



Characterization of Inorganic-Organic Hybrid Solar Cell Electrode Materials and Deposition Methods



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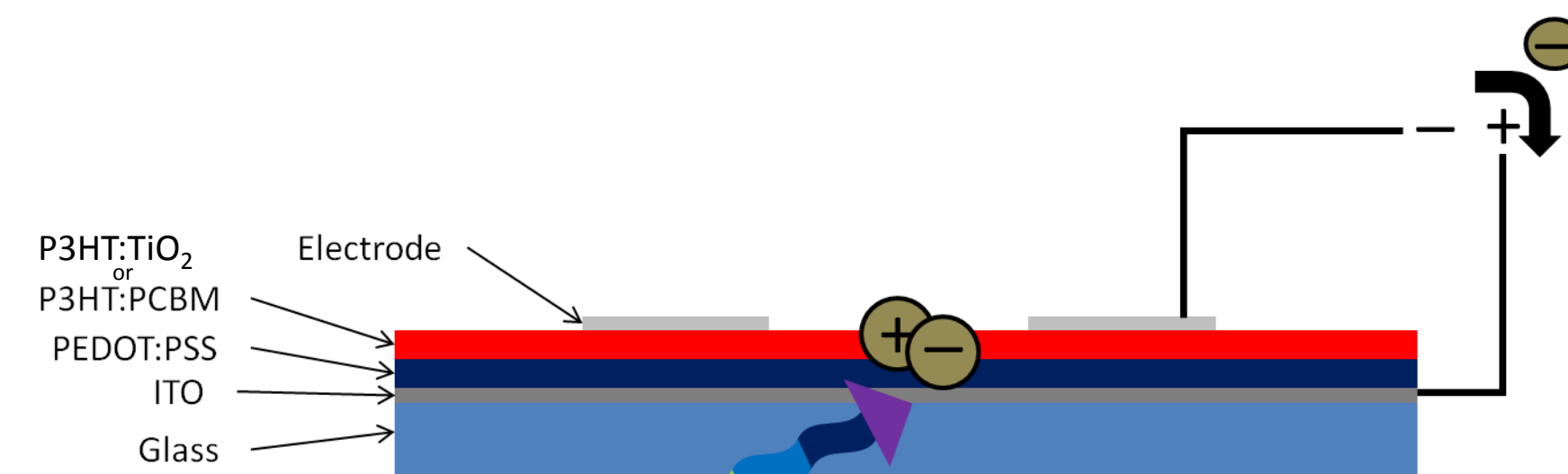
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Introduction

Inorganic-organic hybrid solar cells show promise over traditional silicon based cells due to ease of production and low cost. It is beneficial to analyze potential solar cell electrode materials and deposition processes.

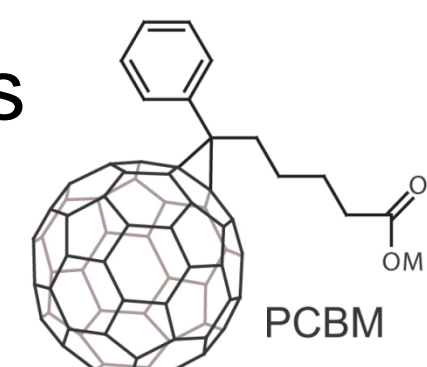
Inkjet-based material printing can reduce costs, waste, and energy inputs. It is known that aluminum evaporation can provide a functioning cell, but little work has been done on printed silver. Solar cell efficiencies were compared between samples with printed silver electrodes and aluminum evaporated electrodes to determine if silver is a viable alternative to aluminum.

Solar Cell Assembly



Indium Tin Oxide (ITO) acts as the bottom electrode.

PEDOT:PSS is a combination of ionomers that increases adhesion and acts as an electron blocking layer.

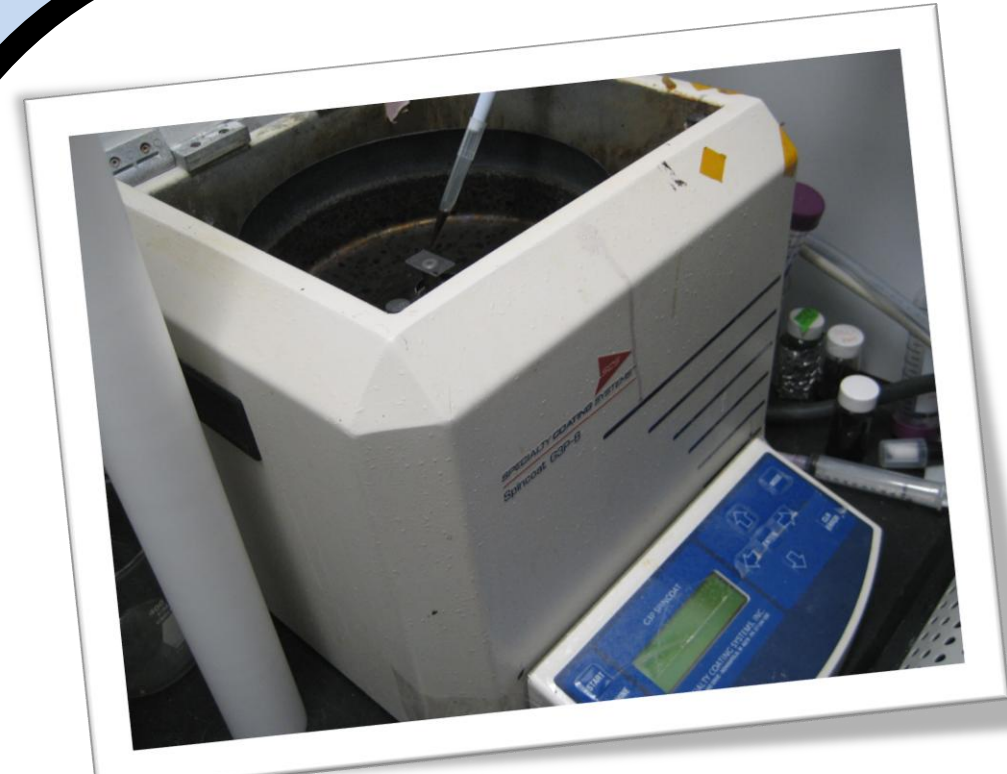


P3HT:PCBM is the active layer of the cell. Light excites the material forming an electron-hole pair called an "exciton".

P3HT:TiO₂ is an alternative active layer. A TiO₂ hole blocking layer (HBL) is required when making this type of cell.

Top electrode is made of either silver or aluminum.

Methods



Spin Coater

- ITO glass obtained from Taiwan University. Glass cut into 7 mm x 20 mm samples with 3 mm ITO strip.
- Spincoat 150 μ L of PEDOT:PSS onto the sample at 5000 rpm for 60 seconds. Samples annealed at 125 $^{\circ}$ C for 10 minutes.

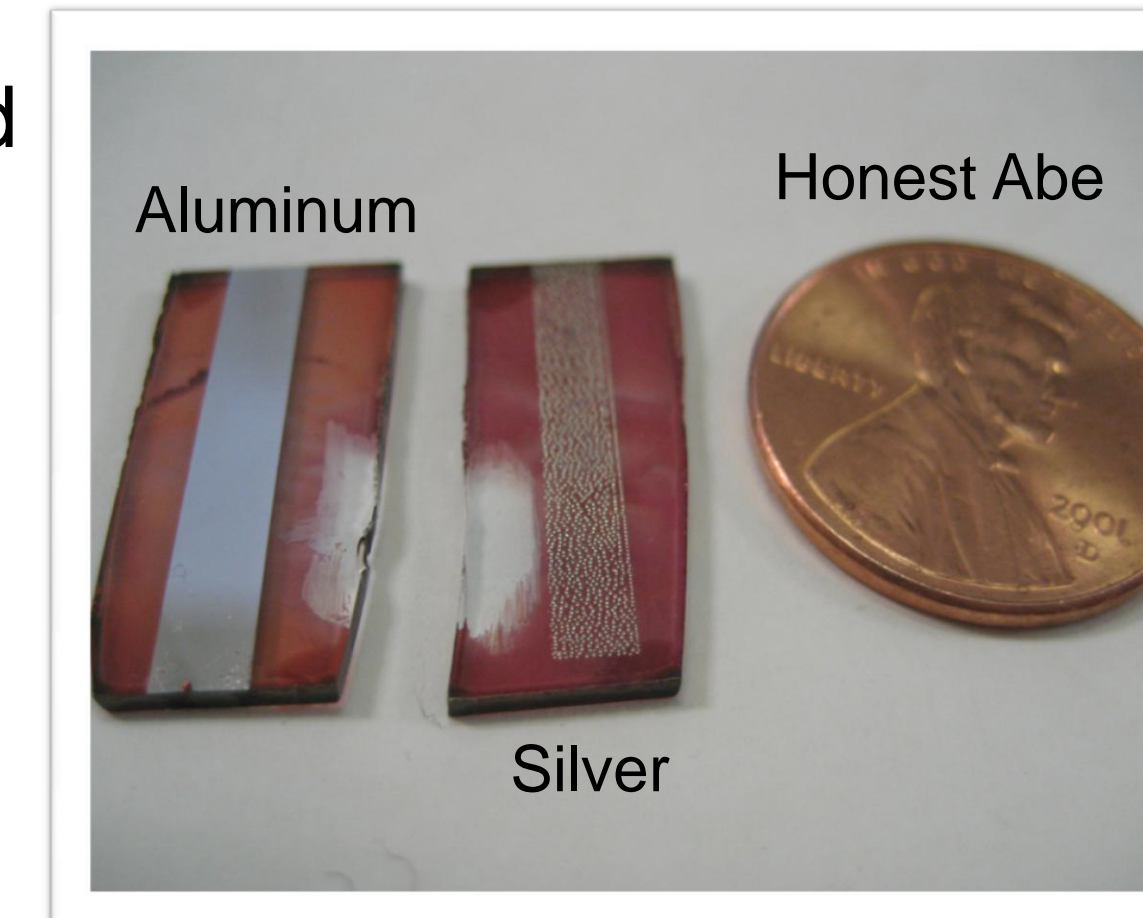
- P3HT:PCBM and P3HT:TiO₂ made up in a 1:1 wt % ratio. Spincoat 150 μ L at 2000 and 1400 rpm respectively for 60 seconds.



Illumination Tower

- Silver electrodes printed with the Dimatix printer then annealed in a vacuum chamber at 140 $^{\circ}$ C.

- Aluminum electrodes flash evaporated in a vacuum then annealed at 140 $^{\circ}$ C.



Completed Solar Cells

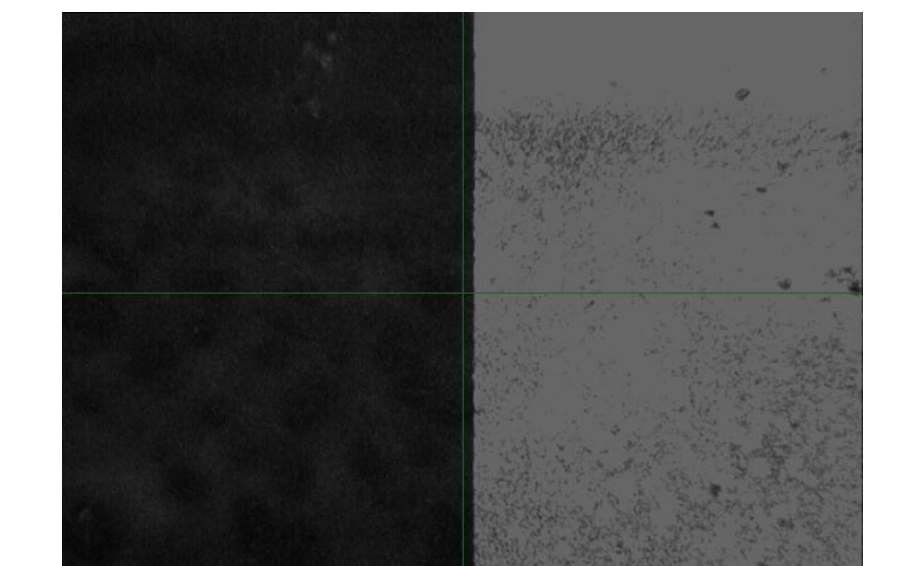
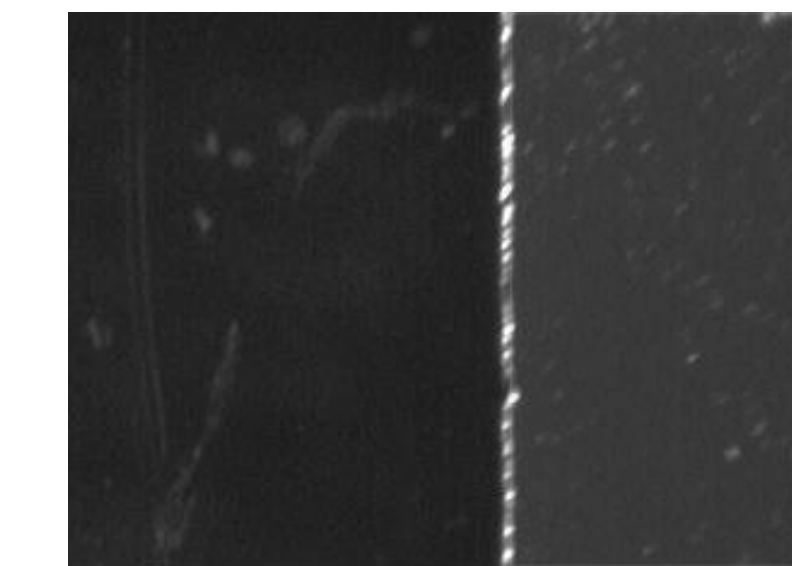
- Illumination testing using the Orion xenon-based sunlight simulator at 100mW/cm².

Electrode Deposition



- Aluminum was deposited onto the P3HT:PCBM active layer via flash evaporation. Evaporation requires a vacuum and high temperatures.

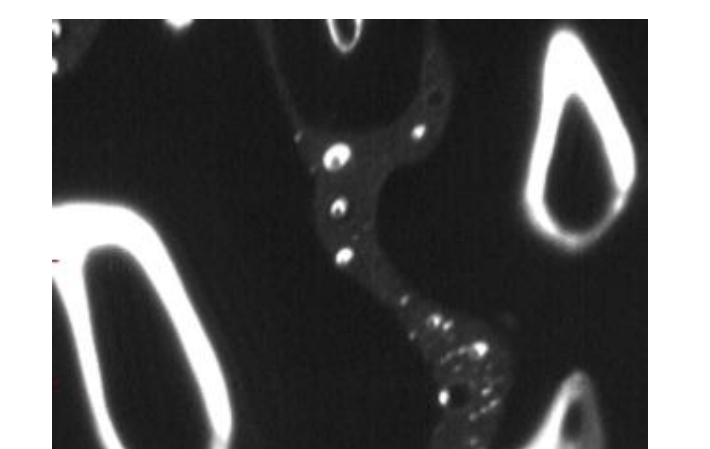
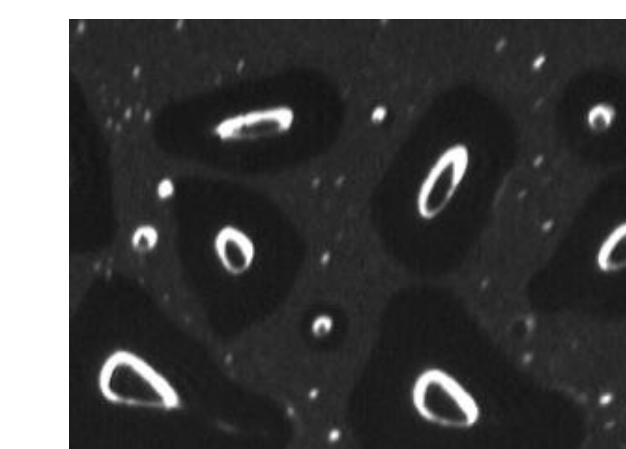
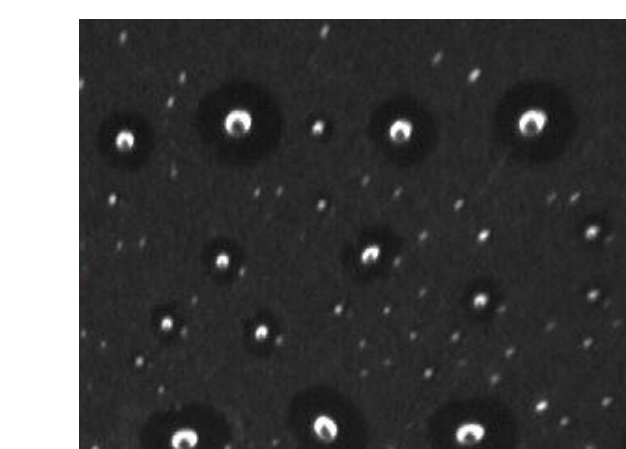
- Evaporation results in a continuous strip along the length of the cell, as shown to the left for P3HT:PCBM cells and right for P3HT:TiO₂ cells.



- Material printers utilize silver inks to deposit electrodes with low energy inputs and short process duration.



- Silver forms discontinuous "islands" when printed onto the P3HT:PCBM. Sequential print passes form larger islands, but yield little increase in continuity.

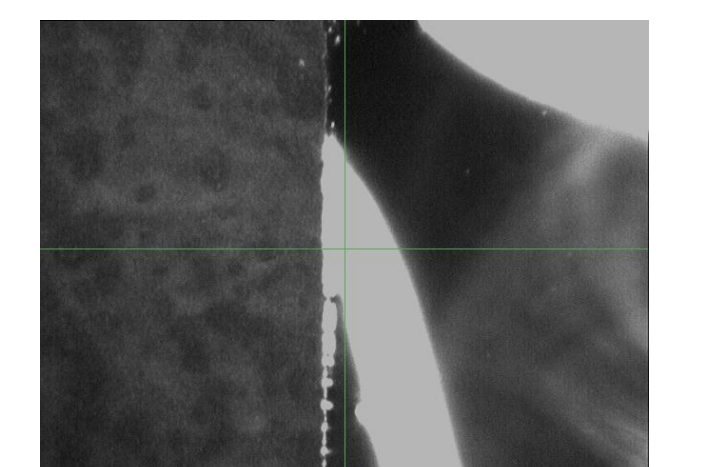


1st print pass

2nd print pass

3rd print pass

- Silver printed onto P3HT:TiO₂ with a TiO₂ HBL layer yields a reflective smooth uniform film.



Results

- Comparison of average solar cell efficiencies are shown below for different electrode types on P3HT:PCBM.

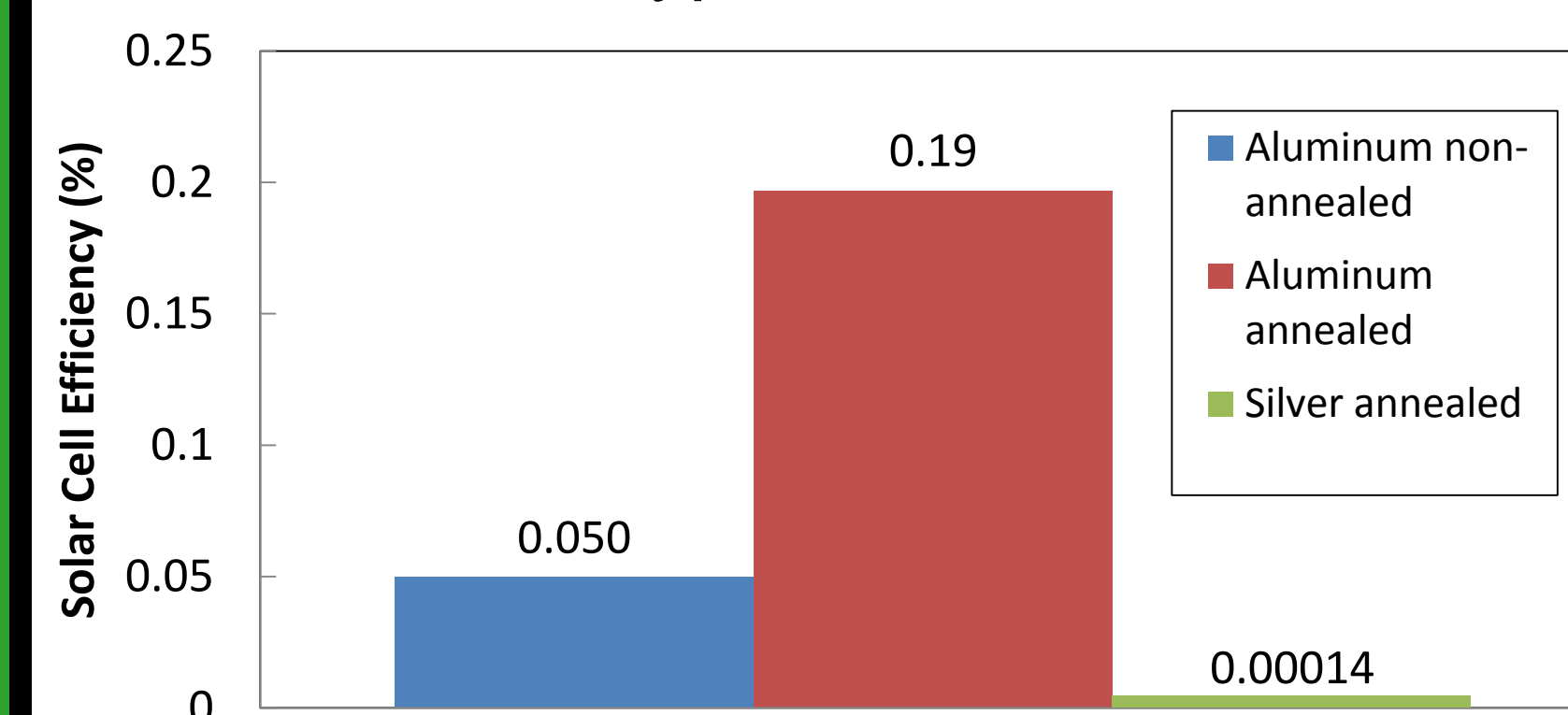


Figure 1 – Annealed solar cells utilizing aluminum electrodes had the highest efficiency for the P3HT:PCBM active layer. Annealing process modifies P3HT morphology, increasing efficiency.

- Solar cell functionality is lost despite having a continuous electrode for printed silver on TiO₂.

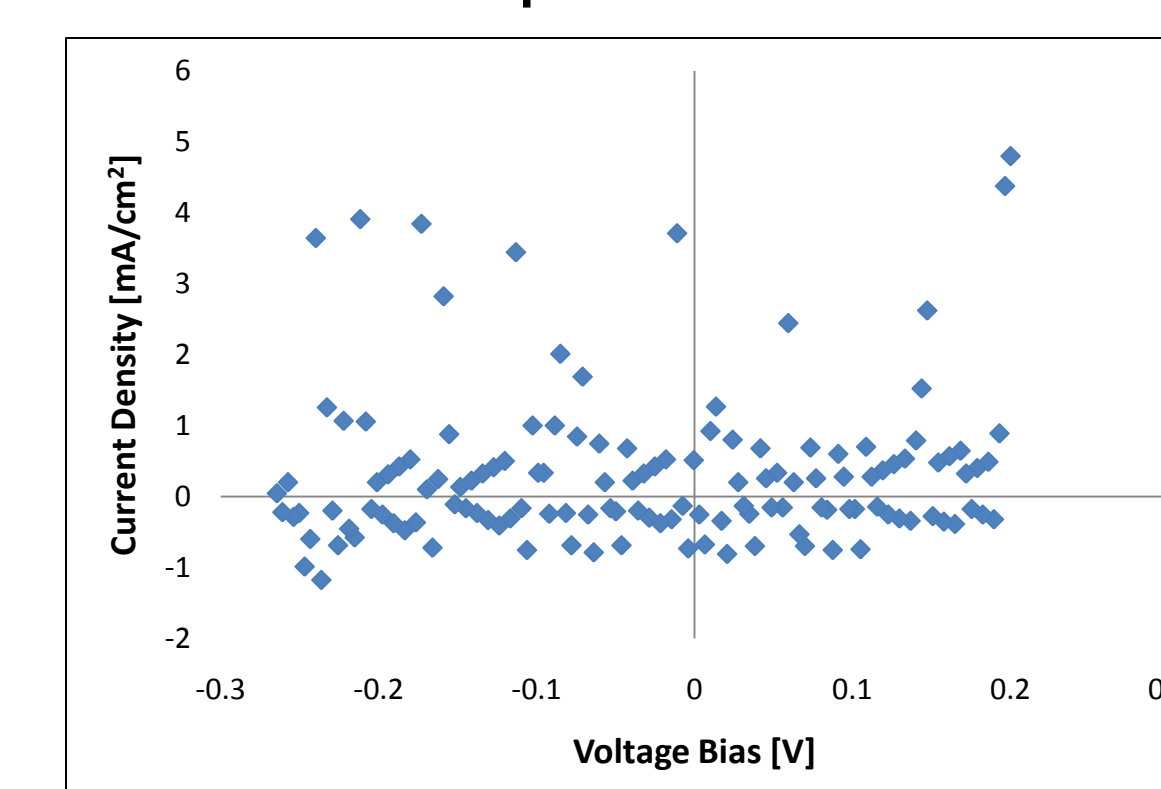


Figure 2 – P3HT:TiO₂ solar cells didn't yield reproducible results. Current density graphs showed patterns of unusable data.